

Electrospark Coatings Based on WC-Co Alloys with Aluminium Oxide and Carbon Additives

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ABSTRACT

The structure and properties of coatings based on WC-Co alloys containing additives of 1% - 5% aluminium oxide and 2.5% - 4.5% carbon were investigated. The coatings had a nanocrystalline structure. Depending on the duration and frequency of the discharge pulses, the ratio between WC and W_2C in the coatings was different. The additives in the hard alloy allowed us increase the microhardness and wear resistance of the coatings by a factor of 2 - 3 in comparison to coatings created using a conventional WC-10%Co alloy.

Keywords: Electrospark Deposition; Coatings; Tungsten Carbide; Carbon; Aluminium Oxide; Wear; Microhardness

1. Introduction

The deposition of wear-resistant electrospark coatings, including nanostructured ones, on the surfaces of machine parts and tools makes it possible to increase their operating lifespan considerably [1-3]. Tungsten carbide-based hard alloys with cobalt binders have become widespread as electrode materials for the fabrication of coatings [4-8]. However, these electrode materials have a high erosion resistance and transfer poorly to the cathode. In addition, cracks often form in the hardened deposited layers under thermal stress when the discharges bombard the cathode. Nanopowder additives of refractory compounds in hard alloys positively affects their microstructure and the tribological characteristics of electrospark coatings [6]. Aluminium oxide, as a simple chemical compound with a high chemical inertness and temperature stability, is an ideal coating component for protection of a hard alloy from the impact of high temperature and chemical wear. An additive of aluminium oxide can also be used as an inhibitor of grain growth [9]. Increasing the Al_2O_3 content in the electrode material from 1% to 5% leads to a decrease in the average grain size from 2.4 to 0.8 μm . Decreasing the grain size of tungsten carbide should lead to an increase in the strength of the hard alloy and also of coatings created on this basis. For example, electrospark coatings created with a nanostructured WC-Co electrode onto an Armco iron were found to exhibit a uniform amorphous structure, higher wear resistance, and reduced friction coefficient as compared to those deposited with conventional microstructured WC-Co

electrodes [7].

On the other hand, it is known that tungsten carbide (WC) decomposes to W_2C and tungsten under the discharge action, and therefore the composition of the coating is different from the initial composition of the treating electrode, that is, the anode [7,8,10]. The decarburation of tungsten carbide reduces the wear resistance of coatings. With increasing power evolving at discharge, the W_2C carbide content of the deposited coating increases. If argon is applied as a protective environment for electrospark processing, the degradation of the tungsten carbide WC decreases but negligibly. Therefore it is necessary to find new ways to diminish the influence of this negative phenomenon.

The major aim of these examinations was to determine the effects of aluminium oxide and carbon additives in the electrode material on the properties of the resulting electrospark coatings.

2. Material and Methods

2.1. Electrode Materials

In our work, electrode materials based on tungsten carbide and cobalt were sintered by the conventional technology of powder metallurgy. Three sets of samples were prepared. The first series of electrodes were of the standard WC-10%Co hard alloy. The second series consisted of the WC-10%Co hard alloy with 1%, 3%, and 5% Al_2O_3 additives. In the third series, all powder mixtures of W, Co, and C were prepared so that the cobalt content was constant (10 wt%) and the carbon concentration was

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varied (2.5 and 4.5 wt%). The powder mixtures were mixed by a PM400 planetary ball mill (Germany) at a rotational velocity of 250 rpm during 30 min with the addition of 5 ml of isopropyl alcohol. The obtained powders were mixed with a solution of rubber in gasoline and then pressed using a Test Press-250M (Russia) with a load of 30 kN. The masses of the samples made for sintering were identical and equalled 13 g. The pressed mixtures were sintered in a Carbolite STF vacuum tube oven (England) at 1400°C during 8 h. The hard alloy sticks used as electrodes had sizes of $40 \times 5 \times 5 \text{ mm}^3$. The mass losses from the sintered samples (induced by plasticizer evaporation and the burning of carbon, tungsten, and cobalt due to the presence of residual oxygen and crystallization water in the initial powders) were 4% on average.

2.2. Coating Deposition and Characterization

The substrates employed were $10 \times 10 \times 5 \text{ mm}^3$ plates of medium-carbon steel 35. The coatings were deposited manually in air by an EMIS (Russia) setup. The inter-electrode voltage during the pulses was maintained at 40 V. The discharge current amplitude was 100 A, while the repetition frequencies of the pulses (f) were 100 or 1000 Hz. The pulse-on length (τ) ranged from 10 to 100 μs , the frequency of the anode vibrator was 100 Hz, and the total processing time was 2 - 20 min. To control the mass transfer from the anode to the cathode, after every 60 s the deposition process was interrupted and the electrodes were weighed separately by an electronic balance (Vibra HTR-220CE) with accuracy up to 10^{-4} g.

The microstructure and chemical composition of the coatings were analysed using an optical microscope and an EVO 40 HV scanning electron microscope (SEM) (Carl Zeiss, Germany) with an INCA Energy 350 energy-dispersion spectrometer (Oxford Instruments, UK). The phase composition of the produced coatings was examined by a DRON-7 X-ray diffractometer (Burevestnik, Russia) equipped with a Cu-K α radiation source in Bragg-Brentano geometry. The coating surfaces were also investigated by AFM microscopy (NT-MTD Company). Microhardness was measured by the Vickers method with a PMT-3M tester (Russia) under a load of 50 N. The abrasive wear resistance of the electrospark coatings was measured by a Calotest friction machine (CSM Instruments, Switzerland) using the "ball-plate" scheme [11]. A ball with a diameter of 30 mm was rotated at a linear velocity of 0.27 m/s. Aluminium oxide powder in solar oil was employed as an abrasive material.

3. Results and Discussion

3.1. Microstructure of Coatings

The microstructures of the created coatings are presented

in **Figure 1**. After etching by 4% nitric acid in ethanol, we observed the ferrite-pearlite structure of the steel substrate; after etching with a mixture of sodium hydroxide and potassium ferricyanide (III), the microstructure is seen in the deposited layer. The white layer contains sections with dispersed iron carbides and aggregations of anode decomposition products. The main part of the surface layer consists of carbide W_2C (the gray area in the upper part of the coating) with small inclusions of carbide WC. **Figure 2** SEM images of the surface and cross-sections of the obtained coatings show the presence of white grains whose elemental composition corresponds to tungsten carbide (WC). Moreover, the quantity of these grains is much larger in the WC-10%Co-4.5%C coating than in the WC-10%Co coating. With increasing carbon content in the electrode, its concentration in the congruous coating increased, since the carbon concentration in the coating on the WC-10%Co-4.5%C sample ranged from 30 to 40 at.%, while that obtained using the standard WC-10%Co electrode was 26 - 31 at.%. The structure of the surface layers deposited by the application of anodes with higher carbon contents was more uniform, although there were some micropores. Thus, the addition of carbon into the electrode materials can increase its content in the coating and prevent decarburization of tungsten carbide due to saturation of the iron present in the coating by additional carbon.

The average thickness of the obtained coatings was 7 - 80 μm (**Table 1**). The thickness of the coating increased with increasing impulse duration.

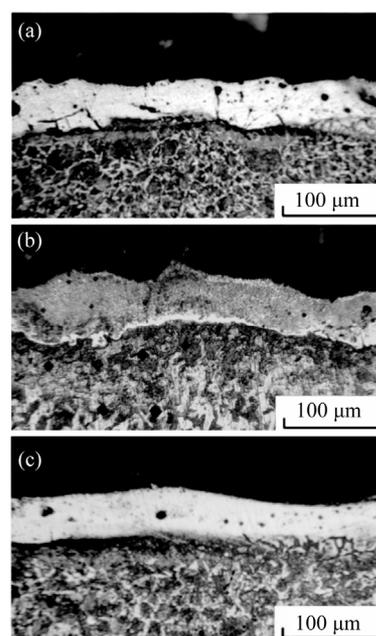


Figure 1. Cross-sectional views of electrospark coatings: (a) WC-10%Co; (b) WC-10%Co-3%Al₂O₃; (c) WC-10%Co-4.5%C.

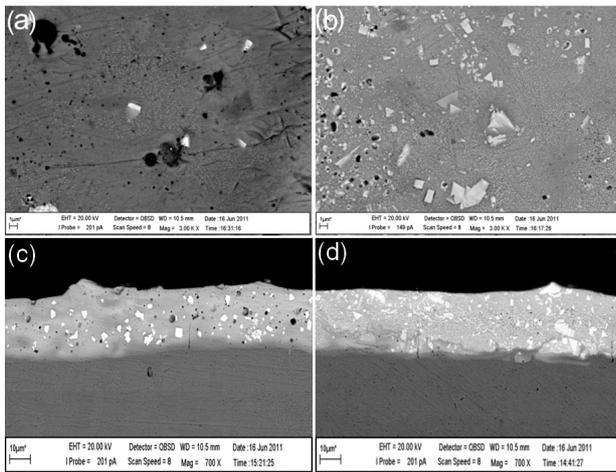


Figure 2. SEM images of the surface (a, b) and cross-section (c, d) of coatings deposited with electrodes WC-10%Co (a, c) and WC-10%Co-4.5%C (b, d).

Table 1. Characteristics of coatings created by electrospark deposition onto steel 35.

Anode	Impulse Length (μs)	Average Thickness (μm)	Average Microhardness (GPa)	Wear Resistance ($\text{N}\cdot\text{m}/\text{cm}^3$)
WC-10%Co	20	7	7	2
	80	20	12	5
WC-10%Co-1%Al ₂ O ₃	40	18	13	15
	80	18	11	40
WC-10%Co-3%Al ₂ O ₃	40	21	11	9
	80	38	13	35
WC-10%Co-5%Al ₂ O ₃	40	25	11	14
	80	37	12	28
WC-10%Co	100	35	16	60
WC-10%Co-2.5%C	100	40	14	73
WC-10%Co-4.5%C	100	37	14	77

Upon metallographic analysis of a transverse section of a cathode with an alloyed surface layer, we found pores and microcracks in the surface layer; the structure was nonuniform over the thickness. The damage to the surface layers increased as the discharge energy evolved. The microcracks were formed due to the difference between the coefficients of thermal expansion for the coating and the steel substrate.

A ferrite-perlite structure was observed on the fragments of steel adjoining the deposited layer. The perlite component is subjected to quenching as a result of the thermal effect and rapid cooling. The section adjacent to the white layer is subject to the thermal action of the discharge and plasma, followed by rapid cooling. As a result, we observed thin layers in which the pearlite component

was quenched. The microhardness was increased in that region (**Figure 1**).

3.2. AFM Investigations

Because the deposited surface layer is relatively rough and scanning probe microscopes cannot be used to study such surfaces, all the samples were given a preliminary assessment on a confocal microscope attached to the SPM. As a result, samples suitable for study using an SPM were selected, and regions on the surface of the alloyed layer where nanostructuring preferentially occurs during electrospark alloying could be identified; they are distinguished by their color (ranging up to blue; **Figure 3(a)**). On the scan obtained by the scanning probe microscope (**Figure 3(b)**), we observed that the nanoparticles were smallest when the electrode consisted of the WC-10%Co-3%Al₂O₃ alloy. The minimum particle size was 30 nm. The figures given in **Figure 3(c)** characterize the adhesion of individual particles. With constant phase composition, increases in this parameter confirm decreases in the size of the nanoparticles.

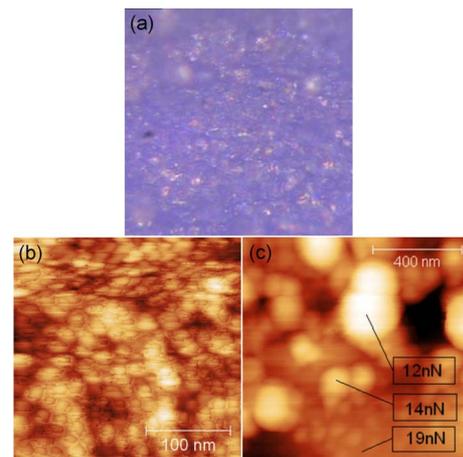


Figure 3. Image of the nanostructured surface of the WC-Co-3%Al₂O₃ coating: (a) View by confocal microscopy; (b) AFM scan; (c) AFM scan with representation of the adhesion of individual particles.

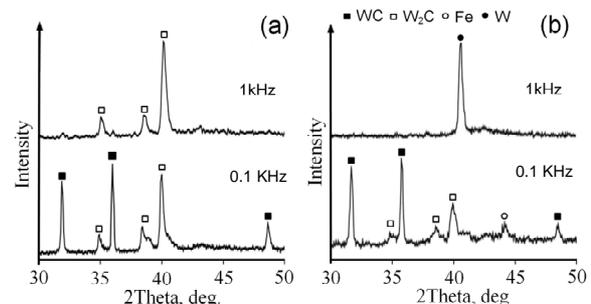


Figure 4. XRD patterns of the coatings obtained by electrospark deposition with (a) WC-10%Co and (b) WC-10%Co-3Al₂O₃ at different frequencies.

3.3. RFA Analysis

Investigation of the phase composition of the coatings (**Figure 4**) showed that the coatings are based on the carbides W_2C with small WC inclusions. The decomposition of tungsten carbide WC to carbide W_2C is observed in local sites as an effect of the electric discharges due to heat liberation and the intensification of the interaction between the electrode material and oxygen in the air. Also, decarburization can occur upon the interaction of WC with the iron substrate, which has been observed for electrospark deposition [7] and HVOF spraying [11]. The compositions of the obtained coatings were different. Such variations started to manifest themselves in an especially pronounced manner as the duration and frequency of the discharge pulses increases. The decarburization of the hard alloy was intense at a higher discharge frequency of 1 kHz. The WC-10%Co coating created at this frequency consisted almost completely of W_2C together with small amounts of WC, in contrast to the same coating deposited at 0.1 kHz. WC carbide dominated over semicarbide W_2C in the coating composition at a frequency of 0.1 kHz. At high frequencies (1 kHz) and short lengths (<20 μ s), the surrounding gas medium exerts no substantial effect.

When electrodes with excess carbon (WC-10%Co-2.5%C or WC-10%Co-4.5%C) were used, tungsten carbide WC dominated in the electrospark coatings. Thus, the addition of carbon into the electrode WC-10%Co material allowed the amount of the WC phase in the deposited layers to increase (**Figure 5**).

3.4. Properties of Electrospark Coatings

Microhardness was measured in a skewed section of samples with coatings. It was established that the microhardness of the WC-10%Co- Al_2O_3 coatings is approximately equal, and greater than that of the steel matrix (2.2 GPa) by a factor of 5 - 6 (**Table 1**). It can be seen that the microhardness of the electrospark coatings deposited decreased with increasing carbon content in the anode materials. This is explained by the fact that there is more WC in these coatings and W_2C is a harder phase as compared to WC but is also brittle. An analysis of the experimental results of wear tests (**Table 1**) shows that the use of the WC-8%Co hard alloy with the aluminium oxide additive as the electrode material allowed us to increase the wear resistance by a factor of four or more in comparison with the steel substrate. Increasing the discharge duration increased the thickness of the coating and its wear resistance.

The excess carbon in the hard alloys increased the wear resistance of the coatings 1.5 fold. The WC-10%Co-2.5%C and WC-10%Co-4.5%C coatings had the highest wear resistance. The wear resistance of the WC-

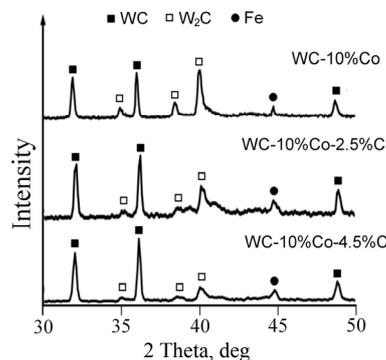


Figure 5. XRD patterns of coatings obtained by electrospark treatment using electrodes with various carbon concentrations at a frequency of 100 Hz.

10%Co-4.5%C coating obtained at a higher discharge frequency of 1 kHz was seven times greater than that of a conventional WC-10%Co sample.

4. Conclusion

We investigated the coatings created by electrospark deposition of WC-10%Co alloys with Al_2O_3 or carbon additives on steel 35. Crystallites of tungsten carbide with a size of 1 - 20 microns were observed in the obtained coatings. The surface layer had a nanostructured construction. With increasing carbon content in the anode material, the amount of tungsten carbide WC increased. The additives of alumina in the electrode materials based on tungsten carbide allowed us to increase the wear resistance of the coatings. If carbon additives were applied, the microhardness and wear resistance became even greater. This is because with increasing carbon concentration in the anode materials, the decarburization of the tungsten carbide decreased.

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